

Final Exam - Solution, Spring Semester 2007

ECSE-6961, Light-emitting diodes and solid-state lighting, Prof. Schubert

Note: In this solution, some of the intermediate steps are left out for brevity.

1. Consider a white-light source consisting of a blue LED ($\lambda = 460$ nm) and a yellow phosphor ($\lambda = 560$ nm). Assume that for every blue photon emitted by the white-light emitter, one yellow photon is being emitted, i.e. 50% of the blue photons are converted to yellow photons.
 - (a) Identify a fundamental energy loss mechanism in this device.
 - (b) What is the maximum attainable power efficiency ($\eta_{\text{power}} = P_{\text{optical}}/P_{\text{electrical}}$) of the device?
 - (c) What is the maximum attainable luminous efficiency of the device?
 - (d) Assume that the blue LED and the yellow phosphor have the following quantum efficiencies: $\eta_{\text{internal}} = \eta_{\text{external}} = 1.0$. Assume further that the device is injected with 1 mA. How many yellow photons are being emitted per second?

(a) Wavelength-conversion loss (also called Stokes-shift loss)

(b) Maximum attainable power efficiency, $\eta_{\text{power}} = \frac{\frac{hc}{\lambda_1} + \frac{hc}{\lambda_2}}{2 \frac{hc}{\lambda_1}} = \frac{1}{2} \left(\frac{\lambda_1 + \lambda_2}{\lambda_2} \right) = 0.91 = 91\%$

where $\lambda_1 = 460$ nm and $\lambda_2 = 560$ nm.

(c) Maximum attainable luminous efficiency, $\eta_{\text{lum}} =$

$$683 \frac{\text{lm}}{\text{W}} \left(\frac{P(\lambda_1) V(\lambda_1) + P(\lambda_2) V(\lambda_2)}{I V} \right)$$

We assume that energy loss occurs only in the phosphor due to the Stokes-shift and no energy is lost inside the semiconductor when converting the input electrical power (IV) into blue photons (hc/λ_1). Therefore,

$$\begin{aligned} \eta_{\text{lum}} &= 683 \frac{\text{lm}}{\text{W}} \left(\frac{\frac{1}{2} \frac{hc}{\lambda_1} V(\lambda_1) + \frac{1}{2} \frac{hc}{\lambda_2} V(\lambda_2)}{\frac{hc}{\lambda_1}} \right) = 683 \frac{\text{lm}}{\text{W}} \frac{1}{2} \left(V(\lambda_1) + \frac{\lambda_1}{\lambda_2} V(\lambda_2) \right) \\ &= 683 \frac{\text{lm}}{\text{W}} \frac{1}{2} \left(0.06 + \frac{460}{560} \times 1 \right) = 301 \text{ lm/W} \end{aligned}$$

(d) Number of yellow photons emitted per second $= \frac{1}{2} \times \frac{I}{e} = \frac{1}{2} \times \frac{1 \times 10^{-3} \text{ A}}{1.6 \times 10^{-19} \text{ C}} =$

$$= 3.125 \times 10^{15} \text{ s}^{-1}$$

2. Consider a white-light source consisting of a UV LED ($\lambda = 400$ nm) and a blue, a green, and a red phosphor emitting at $\lambda = 450$ nm, 550 nm, and 650 nm, respectively. Assume that for every 3 UV photons emitted by the LED, one blue, one green, and one red photon is being emitted.

- What is the maximum attainable power efficiency of the device?
- What is the maximum attainable luminous efficiency of the device?
- Compare the following approaches:
 - White LED made from blue LED and yellow phosphor.
 - White LED made from UV LED and a blue, green, and red phosphor combination.
 Which of the two has the higher power efficiency? Explain the reason in words.
- Which of the two has the higher luminous efficiency? Explain the reason in words.

$$(a) \text{ Maximum attainable power efficiency, } \eta_{\text{power}} = \frac{\frac{hc}{\lambda_1} + \frac{hc}{\lambda_2} + \frac{hc}{\lambda_3}}{3 \frac{hc}{\lambda}} = \frac{1}{3} \lambda \left(\frac{1}{\lambda_1} + \frac{1}{\lambda_2} + \frac{1}{\lambda_3} \right)$$

$$= \frac{400}{3} \left(\frac{1}{450} + \frac{1}{550} + \frac{1}{650} \right) = 74.4\%$$

where $\lambda = 400$ nm, $\lambda_1 = 450$ nm, $\lambda_2 = 550$ nm and $\lambda_3 = 650$ nm.

$$(b) \text{ Maximum attainable luminous efficiency, } \eta_{\text{lum}} =$$

$$= 683 \frac{\text{lm}}{\text{W}} \left(\frac{P(\lambda_1) V(\lambda_1) + P(\lambda_2) V(\lambda_2) + P(\lambda_3) V(\lambda_3)}{I V} \right)$$

We assume that energy loss occurs only in the phosphor due to the Stokes-shift and no energy is lost inside the semiconductor when converting the input electrical power (IV) into UV photons (hc/λ). Therefore,

$$\eta_{\text{lum}} = 683 \frac{\text{lm}}{\text{W}} \left(\frac{\frac{1}{3} \frac{hc}{\lambda_1} V(\lambda_1) + \frac{1}{3} \frac{hc}{\lambda_2} V(\lambda_2) + \frac{1}{3} \frac{hc}{\lambda_3} V(\lambda_3)}{\frac{hc}{\lambda}} \right) =$$

$$= 683 \frac{\text{lm}}{\text{W}} \frac{\lambda}{3} \left(\frac{V(\lambda_1)}{\lambda_1} + \frac{V(\lambda_2)}{\lambda_2} + \frac{V(\lambda_3)}{\lambda_3} \right) = 683 \frac{\text{lm}}{\text{W}} \times \frac{400}{3} \left(\frac{0.05}{450} + \frac{1}{550} + \frac{0.09}{650} \right) =$$

$$= 188.3 \text{ lm/W}$$

- White LED made from blue LED and yellow phosphor has higher power efficiency, because wavelength conversion loss from UV to red is significantly higher thereby reducing the power efficiency.
- White LED made from blue LED and yellow phosphor has higher luminous efficiency, because human eye is significantly less sensitive in the red light region.

3. Consider a 10 km long multimode step-index fiber with a core refractive index of $n = 1.50$ and a cladding refractive index of $n = 1.45$. Assume that the fiber input comes from an LED emitting at 850 nm.
- Calculate the numerical aperture of the fiber.
 - What is the solid angle of a light source that can be coupled into the fiber?
 - Assume that the LED emits 1 mW of optical power isotropically into a hemisphere (solid angle 2π). What is the optical power that can be coupled into the fiber?
 - What optical power exits the fiber?

(a) Numerical Aperture $NA = n_1 \sin \theta_c$

$$\theta_c = \arccos\left(\frac{n_{\text{clad}}}{n_{\text{core}}}\right) = \arccos\left(\frac{1.45}{1.5}\right) = 14.8^\circ$$

$$NA = 1.5 \sin 14.8^\circ = 0.38$$

(b) Solid Angle = $2\pi [1 - \cos(\arcsin(NA))] = 0.47$ steradian

(c) Power coupled = $1 \text{ mW} \times \frac{0.47}{2\pi} = 74.8 \mu\text{W}$

(d) The attenuation loss is about 2 db/km for a silica fiber at 850 nm. Therefore total loss for 10 km will be $2 \text{ db/km} \times 10 \text{ km} = 20 \text{ db}$

$$\text{Loss} = 10 \log(P_{\text{out}}/P_{\text{in}}) \quad \text{or} \quad -20 = 10 \log \frac{P_{\text{out}}}{74.8 \mu\text{W}}$$

$$\text{Solving this equation yields} \quad P_{\text{out}} = 0.748 \mu\text{W}$$

4. This question concerns the design a Distributed Bragg Reflector optimized for a wavelength of 300 nm.
- Select two materials from the table above which would form the DBR with highest reflectivity and least number of periods for 300 nm.
 - Calculate the layer thickness of each material.
 - Which material layer should form the top layer if the ambient is air?

(a) To maximize the refractive-index contrast and also have optical transparency at 300 nm, we select CaF_2 and Si_3N_4 as the two DBR materials.

$$\text{(b) CaF}_2 \text{ layer thickness} = \frac{\lambda}{4 \times \bar{n}_{1,h}} = \frac{300}{4 \times 1.43} = 52.4 \text{ nm}$$

$$\text{Si}_3\text{N}_4 \text{ layer thickness} = \frac{\lambda}{4 \times \bar{n}_{1,h}} = \frac{300}{4 \times 2} = 37.5 \text{ nm}$$

(c) To maximize reflection, material with higher refractive-index contrast with air should be used for the top layer. Therefore, Si_3N_4 should form the top layer.

5. This question concerns colorimetry.
- Deduce the temperature of a black body whose Planckian spectrum has a maximum intensity at 443 nm.
 - Schematically draw a chromaticity diagram and mark the location of the above Planckian black-body radiator. (Hint: Appendix 18.1 on page 312)
 - Can we obtain a light source with the same location on chromaticity diagram as the above Planckian black-body radiator by mixing the blue and yellow sources marked on the chromaticity diagram below?
 - Can we obtain a light source with the same location on chromaticity diagram as the above Planckian black-body radiator by mixing the Blue, Green and Red sources marked on the chromaticity diagram below?
 - Which of the following properties will be the same for the three light sources obtained in (b), (c) and (d)? (i) Spectrum (ii) CRI (iii) x, y color coordinates (iv) u, v color coordinates (v) luminous efficacy.

$$(a) T = \frac{2880 \mu\text{m K}}{\lambda} = \frac{2880 \mu\text{m K}}{443 \times 10^{-3} \mu\text{m}} = 6500 \text{ K}$$

$$(b) x = 0.315, y = 0.327$$

- Yes, because the chromaticity point of the planckian radiator is located on the straight line that can be drawn between the chromaticity locations of the blue and yellow source.
- Yes, because the chromaticity point of the planckian radiator is within the color gamut opened up by the three sources.
- Only the x, y and u, v coordinates will be the same for the light sources.